

LUDWIG-MAXIMILIANS-UNIVERSITÄT MÜNCHEN

## biophysikalische Chemie/molekulare Biophysik Wintersemester 2020/2021

## Solutions to Homework V

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- 1. a) The binding of the first acetylcholine molecule increases the open-to-closed ratio by a factor of  $(1.2 \cdot 10^{-3})$ :  $(5 \cdot 10^{-6}) = 240$ , the binding of the second increases it by a factor of (14):  $(1.2 \cdot 10^{-3}) = 11667$ .
  - b) The free-energy contributions are 14 kJ/mol and 23 kJ/mol, respectively.
- 2. a) From Nernst's Equation, we have:

$$V = \frac{RT}{zF} \ln \left( \frac{C_{extracellular}}{C_{intracellular}} \right)$$

$$V_{Na+} = 152.60 \text{ mV}$$

$$V_{K+} = -243.00 \text{ mV}$$

$$V_{Cl} = -197.96 \text{ mV}$$
 (note that for Cl<sup>-</sup> ion z = -1)

- b) Using the Goldman-Hodgkin-Katz equation, V<sub>M</sub>= -177.43 mV
- c)  $V_{Na+} + V_{K+} + V_{Cl-} = -288.36$  mV This is not the same as b) because to calculate the resting membrane potential, the permeability of the ions also has to be taken into account.
- 3. a) If we consider the electric field across the membrane uniform, its magnitude *E* is given by:

$$E = -\frac{\Delta \phi}{d}$$

where  $\Delta \phi$  is the potential difference between the plates and d is the distance separating the plates.

Using the membrane potential of -70 mV and separation distance of 4.7 nm, we get

$$E = -\frac{-70 \text{ mV}}{4.7 \text{ nm}} = 14.89 \text{ mV/nm} = 1.5 \times 10^5 \text{ V/cm}$$
etric field of this magnitude to two metal electrodes sen

If you applied an electric field of this magnitude to two metal electrodes separated by a 1-cm air gap, they would ionize the air between the electrodes.

b) Using the Nernst equation: 
$$V_{m} = \frac{RT}{zF} \ln \left( \frac{C_{extracellular}}{C_{intracellular}} \right)$$
$$\frac{C_{extracellular}}{C_{intracellular}} = e^{V_{m}} \frac{zF}{RT} = e^{-0.070} \left( \frac{1.96485}{8.314 \cdot 310} \right) = 0.07276 = \frac{1}{13.74}$$

4. a) Given that

$$\frac{dn}{dt} = \alpha_n(V)(1-n) - \beta_n(V)n$$

Initially, with  $V_m = 0$ , the state variable n has a steadystate value (i.e., when dn/dt = 0) given by:

$$n_{\infty}(0) = \frac{\alpha_n(0)}{\alpha_n(0) + \beta_n(0)}$$

When  $V_m$  is clamped to a new level  $V_c$ , the gating variable n will eventually reach a new steady state value given by:

$$n_{\infty}(V_c) = \frac{\alpha_n(V_c)}{\alpha_n(V_c) + \beta_n(V_c)}$$

The solution to the first equation that satisfies these two boundary conditions is a simple exponential of the form:

$$n(t) = n_{\infty}(V_c) - (n_{\infty}(V_c) - n_0(0))e^{-t/\tau_n(V_c)}$$

This equation describes the time course of n in response to a step change in command voltage

b) The original values from the paper are:

$$\begin{split} \alpha_n &= 0.01 \; (V+10) \bigg/ \bigg( \exp \frac{V+10}{10} - 1 \bigg), \\ \beta_n &= 0.125 \; \exp \; (V/80), \\ \alpha_m &= 0.1 \; (V+25) \bigg/ \bigg( \exp \frac{V+25}{10} - 1 \bigg), \\ \beta_m &= 4 \; \exp \; (V/18), \\ \alpha_h &= 0.07 \; \exp \; (V/20), \\ \beta_h &= 1 \bigg/ \bigg( \exp \frac{V+30}{10} + 1 \bigg). \end{split}$$

Rate Constants	A	В	С	D	F	Н
$\alpha_{\rm n}$	0.1	0.01	-1	10	10	1
$\beta_n$	0.125	0	0	0	80	1
$\alpha_{\mathrm{m}}$	2.5	0.1	-1	25	10	1
$\beta_{m}$	4	0	0	0	18	1
$lpha_{ m h}$	0.07	0	0	0	20	1
$\beta_h$	1	0	1	30	10	1